Selective Perfluoro- and Polyfluoroarylation of Meldrum's Acid

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Supporting Information

ABSTRACT: This work describes the facile and monoselective per- and polyfluoroarylation of Meldrum's acid to generate a versatile synthon for highly fluorinated α -phenyl acetic acid derivatives, which provide straightforward access to fluorinated building blocks. The reaction takes place quickly, and most products were isolated without the need for chromatography. Importantly, this method provides an alternative strategy to access α -arylated Meldrum's acids,



which avoids the need for aryl-Pb(IV) salts or diaryliodonium salts. Furthermore, we demonstrate the synthetic versatility and utility of the Meldrum's acid products by subjecting our products to several derivatizations of the Meldrum's acid products as well as photocatalytic hydrodefluorination.

INTRODUCTION

Meldrum's acid was initially discovered in 1908 by Meldrum;¹ however, the structure was misidentified for the next 40 years until it was correctly assigned by Davidson.² Because of its anomalous acidity ($pK_a = 7.3$ in DMSO),³ Meldrum's acid has a long and rich history as an activated nucleophile,⁴ and, contrary to its malonate cousin, it can be hydrolyzed easily under acidic conditions, which can allow for facile elaboration not possible with malonates. While strategies for the selective alkylation of Meldrum's acid have been well developed such as the reductive alkylation of aldehydes,⁵ coupling and reduction of carboxylic acids,⁶ addition to Michael acceptors,⁷ substitution of Mitsunobu reagents,⁸ substitution of alkyl halides,⁷ and the addition to cationic metal allyls,⁹the corresponding arylation is far less developed.

In contrast to alkylation of Meldrum's acid, very few methods exist for the direct α -arylation of Meldrum's acid.¹⁰ In fact, most commonly this motif is achieved via acetal or ketal formation of an α -phenyl malonic acid or its ketene derivative.¹¹ A more direct approach would be to α -arylate an already existing Meldrum's acid unit. Toward this goal, Chen and Stang¹² have shown that diaryliodonium salts can afford direct arylation of Meldrum's acid (eq 1, Scheme 1). Furthermore, Pinhey¹³ has shown that aryl-leadtriacetates undergo facile coupling with Meldrum's acid (eq 2, Scheme 1).

Comparison to Previous Arylations of Meldrum's Acid. Unfortunately, neither of these methods allows for selective monoarylation as the reaction always leads to the fully quaternerized product. Thus, methods that allow for the selective monoarylation of Meldrum's acid are needed to be able to access the Meldrum's acid that possesses a tertiary center or unsymmetric quaternerized center.

Recently, our group has taken an interest in increasing the number of fluorinated arene building blocks as well as methods for their further elaboration. A method for direct per(poly)- fluoroarylation of Meldrum's acid would be particularly ideal given the versatility of Meldrum's acid.

Given perfluoroarenes are known to readily undergo nucleophilic aromatic substitution (S_NAr) ,¹⁴ we were surprised to find that the addition of Meldrum's acid had never been reported (eq 3, Scheme 1), and we set about to develop this reaction.¹⁵ We hoped to determine whether Meldrum's acid is a competent nucleophile for S_NAr reactions of highly fluorinated arenes. Herein, we report conditions that allow for clean, rapid, and operationally simple monoarylation of Meldrum's acid arenes. Furthermore, we demonstrate that these adducts are easily functionalizable and undergo typical reactions expected of Meldrum's acid.

RESULTS AND DISCUSSION

We initiated our investigation using pentafluoropyridine and 1,5-dioxaspiro [5.5] undecane-2,4-dione (MA) along with diisopropylethylamine (DIPEA) in acetonitrile (Table 1). In this analogue, the normal [2,2] dimethyl group has been replaced with a cyclohexyl group. Consequently, it is more soluble¹⁶ in most organic solvents and displays a greater hydrolytic stability¹⁷ than simple Meldrum's acid (2,2-dimethyl-1,3dioxane-4,6-dione), which can be prone to hydrolysis over prolonged time periods. With routine optimization, several trends became clear. First, polar aprotic solvents worked well (entries 1-3), while protic- (entry 4), halogenated- (entry 5), aromatic- (entry 6), and ethereal-solvents (entry 7) were found to be inferior. Acetonitrile was used for further studies because its comparatively greater volatility facilitated isolation. Additionally, both triethylamine (entry 8) and K_2CO_3 (entry 9) were screened as bases. While triethylamine gave conversions similar to those of the standard reaction (entry 1), it also

Received: September 11, 2014 Published: October 1, 2014

Scheme 1. Previous Direct Arylations of Meldrum's Acid

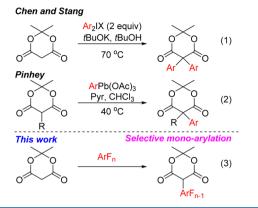
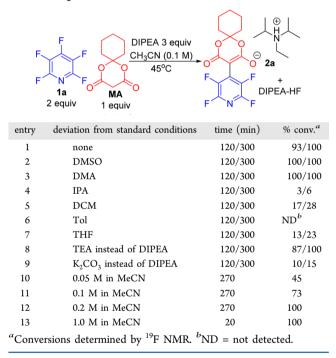


Table 1. Optimization of Reaction Conditions



yielded an undesirable *N*-arylated side-product.¹⁸ Thus, DIPEA was used for further studies. Finally, the effect of concentration was briefly examined (entries 10-13). The rate of the reaction displayed a dependency on the concentration, and thus the reactions were run at 1 M concentration to shorten the reaction times.

With conditions in hand that provided product with remarkable speed and were essentially free of unwanted side products, we next sought to develop workup conditions that would allow us to isolate the DIPEA-adduct salt from left-over perfluoroarene, DIPEA, and DIPEA-HF salt. Fortuitously, we were able to accomplish this through a series of evaporations, solvent changes, and washings, which resulted in pure products.¹⁹ The workups allow rapid isolation in high yield with no need for column chromatography, which should facilitate its implementation on a larger scale.²⁰

With reaction and workup conditions in hand, we began to evaluate the scope of the reaction. In general, this reaction works extremely well for fluoroarenes that possess an activating functional group (i.e., an electron-withdrawing group) (Table 2).

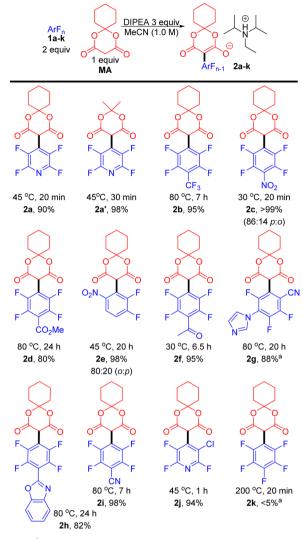


Table 2. Addition of Unsubstituted MA to Fluorinated

Arenes

 $^{a}\mu$ -wave heating.

Specifically, the reaction worked well for pyridines (1a, 1a', and 1j), nitro-substituted substrates (1c and 1e), ketones (1f), nitriles (1g and 1i), and esters (1d), as well as trifluoromethyl groups (1b) and heterocycle substituted substrates (1g and 1h). Only in the case of the nitroarenes 1c and 1e were regioisomers observed. Selective addition to the C-4 (fluorine bearing) of 3-chloro-2,4,5-trifluoropyridine (1j) is consistent with the S_NAr mechanism. The absolute control of monoarylation versus diarylation is remarkable considering that the previous methods were unable to selectively monoarylate.^{12,13}

Unfortunately, the reaction would not work for simple hexafluorobenzene even at elevated temperatures (200 °C) where it remained mostly unchanged. It did, however, demonstrate the robustness of MA at high temperatures under basic conditions, as no decomposition products could be detected. Speculating that the lack of reactivity of hexafluor-obenzene (1k) could be attributed to an insurmountable reaction barrier, we reasoned that it might be overcome by increasing the reactivity of the nucleophile. To accomplish this, we turned to the malonate ester, which is significantly more basic (diethyl malonate $pK_a = 16.4$ in DMSO)²¹ and correspondingly more nucleophile.²² We were pleased to find

The Journal of Organic Chemistry

that use of the malonate did indeed lead to clean substitution of even the least activated of the perfluoroarenes (Table 3). While we would have liked to have used Meldrum's acid as the nucleophile due to its synthetic versatility, ultimately we were able to obtain the 1,3-diester motif using diethyl malonate for hexafluorobenzene (3k), decafluorobiphenyl (3l), and perfluoronaphthalene (3m).^{15a,23}

Next, we sought to fully quaternerize the remaining activated methine via alkylation. However, after several failed attempts to methylate 2a via standard alkylation techniques, we were forced to reevaluate our strategy. The lack of alkylation could be due to either steric or electronic inhibition (or alternatively a combination of both). However, we suspected that the additional stabilization of the MA-carbanion by the perfluoroarene significantly reduced the nucleophilicity of the carbanion and that the rate of alkylation (and subsequent arylation) was slow due more to the electronics than the steric inhibition of the transition state. Steric inhibition seems less likely given that there are numerous examples of quaternerized Meldrum's acid compounds that were achieved via alkylation,²⁴ although admittedly there are no examples of alkylations of α -(2,6-disubstituted aryl)-Meldrum's acid. Nonetheless, we were forced to reconsider our planned route to the fully quaternerized carbons.

In an attempt to quaternerize the MA, we next reversed the order of events. Using α -alkylated Meldrum's acid derivatives, we were able to facilitate the S_NAr reaction to arrive at the desired fully substituted Meldrum's acid derivatives (Table 4). We were pleased to find that the additions took place smoothly with no alterations from the initial conditions despite the fact that the reaction forms a quaternary center, hinting that the inability to alkylate **2a** is an electronic rather than a steric issue. As can be seen from Table 4, the yields range from good to excellent. To the first approximation, the reactions take place with rates similar to those of the corresponding perfluoroarylation (i.e., Table 2).

Finally, we wanted to demonstrate the utility of the MAadducts as highly functionalizable building blocks (Scheme 2). While our MA-adducts were stable under basic conditions, under acidic conditions they undergo facile hydrolysis²⁵ and nucleophilic addition.²⁶ Under acidic conditions, nucleophilic attack of the carbonyl group and ring opening take place along with decarboxylation. We have found that the decarboxylation is significantly accelerated by the presence of the perfluoroaryl group such that it takes place readily at methanolic reflux (and

 Table 3. Addition of Diethylmalonate to Unactivated

 Perfluoroarenes

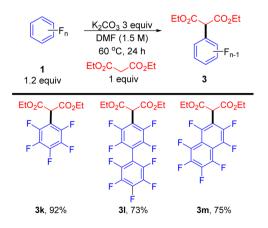
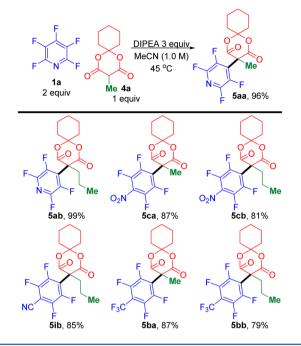
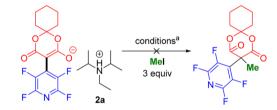


Table 4. Addition of α -Alkylated MAs to Perfluoroarenes



Scheme 2. Failed Direct Alkylation of the Adducts^a



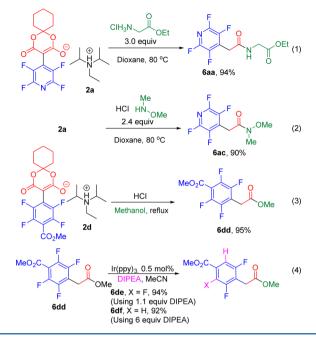
a(a) See the Supporting Information for details on conditions tried.

even at room temperature)¹⁹ rather than temperatures well over 100 °C²⁷ typical of the relevant diacids. Glycine ethyl ester underwent smooth acylation to afford the *N*-addition product (**6aa**, eq 1, Scheme 3). The Weinreb amide, which is a useful synthetic intermediate for the formation of ketones, was easily formed by the addition of its HCl salt with gentle heating (**6ac**, eq 2). Methanolysis of MA-adduct **2d** (eq 3) cleanly formed the dimethyl ester, which we subjected to photocatalytic hydrodefluorination (HDF) conditions (eq 4) recently developed in our lab.²⁸ By controlling the equivalents of reductant (the aliphatic amine) and the reaction time, selective formation of both the mono-HDF product (**6de**) and di-HDF (**6df**) was accomplished, which allowed access to a di- and trifluorinated arene that would be challenging to access by existing methods.

CONCLUSIONS

We have reported conditions that facilitate the addition of Meldrum's acid to perfluoro- and polyfluoroarenes to arrive at synthetically versatile highly fluorinated building blocks. Furthermore, we have shown that we can take advantage of the relatively acidic nature of the products to prevent over arylation and to facilitate isolation. We also have shown that fully quaternerized carbons can be synthesized by reversing the order of events, which leverages the wealth of methods that exist for selective alkylation to circumvent the problematic alkylation. Given the increasing demand for fluorine incorpo-

Scheme 3. Utility of the MA-Adducts

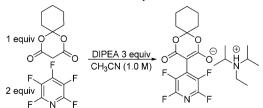


ration into molecules of disparate fields, we suspect that this simple method for construction of fluorinated building blocks will be of value for a broad spectrum of chemists.

EXPERIMENTAL SECTION

All reagents were obtained from commercial suppliers and used without further purification unless otherwise noted. Acetonitrile (CH₃CN) was dried over molecular sieves. Photocatalyst tris-(2phenylpyridinato- C^2 , N)iridium(III), 99% (purity) (Ir(ppy)₃), was obtained from a chemical supplier. Methyl-2,3,4,5,6-pentafluorobenzoate (1d)²⁸ 2-(perfluorophenyl)benzo[d]oxazole (1h),²⁸ 2,3,5,6-tetrafluoro-4-(1H-imidazol-1-yl)benzonitrile²⁹ (1g), and cyclohexyl-Meldrum's acid, MA,¹⁶ were synthesized according to literature procedures. NMR spectra were obtained on 400 MHz spectrometers. ¹H and ¹³C NMR chemical shifts are reported in ppm relative to the residual protio solvent peak (¹H, ¹³C). Melting points were reported uncorrected. HRMS data were obtained using a LTQ (linear trap quadrupole) Orbitrap XL mass spectrometer. Isolations were carried out using an automated chromatography system using normal phase silica (4, 12, 24, or 40 g) with product detection at 254, 280 nm and using an evaporative light scattering detector (ELSD). Substrate syntheses reactions were monitored by thin layer chromatography (TLC), with UV254, glass backed, 250 μ m, and were visualized with ultraviolet light or potassium permanganate.

General Procedure A: Fluoroarylation of Unsubstituted-MA.



For the monitoring purposes, these reactions were carried out in NMR tubes and monitored via ¹⁹F NMR. An NMR tube was charged with cyclohexyl-Meldrum's acid (1 equiv), then CH₃CN (1.0 M), followed by N_iN -diisopropylethylamine (3 equiv), and fluorinated starting material **1a**–**k** (1.2 or 2 equiv) were added, and a sealed glass capillary containing C₆D₆ was placed in the NMR tube for locking purposes. The NMR tube then was placed in a sand bath and maintained at the appropriate temperature (30, 45, or 80 °C). The NMR tube was

occasionally removed from the sand bath, and the progress was checked by $^{19}\mathrm{F}$ NMR.

General Workup Method B for the Isolation of the Fluoroarylated-MA Adduct (for Volatile Starting Materials). After reaction completion as judged by ¹⁹F NMR, CH₃CN and excess volatile starting material were removed via rotavap, and the residue was dissolved in DCM (2 mL) and washed with deionized water (2 mL) to selectively remove DIPEA-HF salt. The organic layer was dried with anhydrous MgSO₄, filtered, and concentrated in vacuo to yield the title compound.

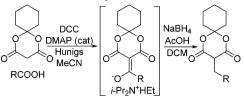
General Workup Method C for the Isolation of the Fluoroarylated-MA Adduct (for Nonvolatile Per- and Polyfluoroarene Starting Materials). In the case of less volatile fluoroarene starting materials, purification of the crude material was done by using the following purification method. After reaction completion as judged by ¹⁹F NMR, CH₃CN was removed via rotavap, and the residue was dissolved in DCM (2 mL) and washed with deionized water (2 mL) to remove DIPEA-HF salt. The organic layer was dried with anhydrous MgSO₄, filtered, and concentrated in vacuo. The compound was purified via trituration. In a borosilicate test tube, the sample was dissolved in a minimal amount (~0.5 mL) of ethyl acetate, and pentane (~5.0 mL) was slowly added along the wall of the test tube resulting in the formation of two layers. The test tube then was capped with a rubber septum and allowed to sit overnight, which resulted in the selective dissolution of the excess fluoroarene starting material (typically 1.2 equiv used) and the ethyl acetate into the pentane layer, leaving behind the product as a solid or oil. The product was then separated by simple decantation of the pentane layer, which contained the excess starting material. Finally, the product was placed in a vacuum to remove any left-over pentane to yield the title compound.

General Procedure D: Fluoroarylation of Diethyl Malonate. These reactions were carried out in 0.5–2.0 mL microwave vials. A microwave vial was charged with diethyl malonate (1 equiv), DMF (1.6 M), fluorinated starting material (1.2 equiv), and K_2CO_3 (3.0 equiv), and then a magnetic stir bar was added. The microwave vial was placed in a sand bath and maintained at 60 °C. The reaction was monitored by ¹⁹F NMR by pulling out small aliquots using a syringe and a needle. After the completion of the reaction, the residue was dissolved in DCM (~4 × the vol of DMF) and washed with deionized water (3 × 5 mL) to remove K_2CO_3 and DMF. The organic layer was dried with anhydrous MgSO₄, filtered, and concentrated in vacuo. The crude product was purified by normal phase chromatography with hexanes and ethyl acetate.

General Procedure E: Fluoroarylation of Substituted-MA. For monitoring purposes, these reactions were carried out in NMR tubes and monitored via ¹⁹F NMR. An NMR tube was charged with alkylated-Meldrum's acid (1 equiv), and then CH₃CN (1.0 M), fluorinated starting material **1a**–**c** (2 equiv), and *N*,*N*-diisopropylethylamine (3 equiv) were added, and a sealed glass capillary containing C_6D_6 was placed in the NMR tube for locking purposes. The NMR tube then was placed in a sand bath and maintained at the appropriate temperature (30, 45, or 80 °C). The reaction was monitored by ¹⁹F NMR.

General Procedure F: Photocatalytic Hydrodefluorination Reaction. This reaction was carried out following the literature procedure.¹ In an NMR tube capped with NMR septum was charged tris-(2- phenylpyridinato- C^2 ,N) iridium(III) (Ir(ppy)₃) (1 mL of a 0.5 mM solution in MeCN). Fluorinated starting material **6dd** (1.0 equiv) and *N*,*N*-diisopropylethylamine (1.1 or 6.0 equiv) were added, and a sealed glass capillary containing C_6D_6 was added for locking purposes before degassing. The reaction then was degassed through the septum via Ar bubbling for 5–10 min at 0 °C (to avoid evaporation of *N*,*N*diisopropylethylamine). The NMR tube was placed in a light bath, and the lower portion of the tube was submerged under the water bath, which was maintained at 45 °C. The reaction was periodically monitored by ¹⁹F NMR. After the reaction completion, CH₃CN was removed via rotavap, and the residue was treated with deionized water (2 mL) and extracted with EtOAc (5 × 1 mL). The combined organic portions were dried with anhydrous MgSO₄, filtered, concentrated in vacuo, and purified by normal phase chromatography.

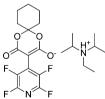




These reactions were carried out in a one-pot, two-step procedure.³⁰ In a 100 mL round-bottom flask, the corresponding carboxylic acid (1.0 equiv) and MeCN (0.2 M) were added followed by cyclohexyl-Meldrum's acid (1.1 equiv). To the stirring mixture was added DMAP (4-dimethylaminopyridine) (0.1 equiv) followed by the dropwise addition of N,N-diisopropylethylamine (2.15 equiv). Next, DCC (N,N'-dicyclohexylcarbodiimide) (1.1 equiv) was added, and the reaction was stirred at room temperature for 14 h. MeCN then was evaporated in vacuo, and the crude material was taken onto the next step without further purification. In the second step, the crude material was dissolved in DCM (0.2 M), cooled to 0 °C, and acetic acid (10 equiv) was added followed by the portion-wise addition of NaBH₄ (2.5 equiv). The reaction mixture was allowed to warm to room temperature and stirred overnight. The reaction was then diluted with water (20 mL), acidified with 1 M HCl. The organic layer was separated, dried over anhydrous MgSO₄, and concentrated in vacuo. The resultant crude residue was purified by automated flash chromatography to give the product.

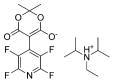
General Procedure H: Aminolysis Reaction of the Fluoroarylated-MA Adduct. For monitoring purposes, these reactions were carried out in NMR tubes and monitored via ¹⁹F NMR. An NMR tube charged with 2a (1.0 equiv) and the amine-HCl salt (2.4 or 3.0 equiv) and dioxane (0.1 M) was added, and sealed glass capillary containing C_6D_6 was placed in the NMR tube for locking purposes. The NMR tube then was placed in a sand bath and maintained at 80 °C. The reaction was periodically monitored by ¹⁹F NMR. After the completion of the reaction, the residue was dissolved in DCM (2 mL) and washed with deionized water (3 × 5 mL) to remove dioxane. The organic layer was dried with anhydrous MgSO₄, filtered, and concentrated in vacuo to obtain the title compound.

Synthesis of *N*-Ethyl-*N*-isopropylpropan-2-aminium 4-Oxo-3-(perfluoropyridin-4-yl)-1,5-dioxaspiro[5.5]undec-2-en-2olate (2a).



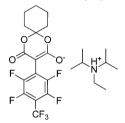
The general procedure A was followed using pentafluoropyridine (220 μ L, 2.0 mmol, 2.0 equiv), cyclohexyl-Meldrum's acid, MA (184 mg, 1.0 mmol, 1.0 equiv), *N*,*N*-diisopropylethylamine (522 μ L, 3.0 mmol, 3.0 equiv), and CH₃CN (1.0 mL). After the completion of the reaction, the workup method B was used to isolate **2a** in 90% yield as a light yellow solid (416.0 mg, 0.9 mmol). ¹⁹F NMR (376 MHz, CDCl₃): δ –96.5 to –96.8 (m, 2F), –138.6 to –138.9 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 3.64–3.50 (m, 2H), 3.04 (qd, *J* = 7.4, 4.0 Hz, 2H), 2.08–1.98 (m, 4H), 1.65 (p, *J* = 6.6 Hz, 4H), 1.50–1.32 (m, 17H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 164.2, 144.8–141.9 (m), 142.1–138.5 (m), 131.9 (ddd, *J* = 20.1, 16.8, 3.5 Hz), 103.1, 67.7, 54.4, 42.7, 34.8, 25.0, 22.7, 18.4, 17.0, 12.4. mp 138–140 °C. FT-IR (neat): cm⁻¹ 2935, 1605, 1254, 1368, 1466. HRMS (ESI) calcd, C₁₄H₁₀F₄NO₄⁻ 332.0551; observed, 332.0532.

Synthesis of *N*-Ethyl-*N*-isopropylpropan-2-aminium 2,2-Dimethyl-4-oxo-5-(perfluoropyridin-4-yl)-4*H*-1,3-dioxin-6olate (2a').



The general procedure A was followed using pentafluoropyridine (110 μ L, 1.0 mmol, 2.0 equiv), simple-Meldrum's acid (2,2-dimethyl-1,3-dioxane-4,6-dione) (72.0 mg, 0.5 mmol, 1.0 equiv), N,N-diisopropylethylamine (261 μ L, 1.5 mmol, 3.0 equiv), and CH₃CN (0.5 mL). After the completion of the reaction, the workup method B was used to isolate **2**a' in 98% yield as a white solid (207.0 mg, 0.49 mmol). ¹⁹F NMR (376 MHz, CDCl₃): δ –95.6 to –95.9 (m, 2F), –138.2 to –138.4 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 3.47 (hept, *J* = 13.0, 6.5 Hz, 2H), 3.04–2.90 (m, 2H), 1.60 (s, 6H), 1.31–1.21 (m, 15H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 164.5, 144.9–141.9 (m), 142.0–138.9 (m), 131.8 (tt, *J* = 16.6, 3.4 Hz), 102.5, 67.6 (d, *J* = 4.6 Hz), 54.4, 42.7, 26.0, 18.4, 17.1, 12.4 mp 102–104 °C. FT-IR (neat): cm⁻¹ 2992, 2958, 1600, 1266. HRMS (ESI) calcd, C₁₁H₆F₄NO₄⁻ 292.0238; observed, 292.0217.

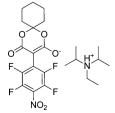
Synthesis of *N*-Ethyl-*N*-isopropylpropan-2-aminium 4-Oxo-3-(2,3,5,6-tetrafluoro-4-(trifluoromethyl)phenyl)-1,5dioxaspiro[5.5]undec-2-en-2-olate (2b).

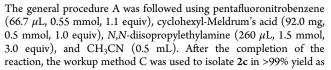


The general procedure A was followed using octafluorotoluene (283 μ L, 2.0 mmol, 2.0 equiv), cyclohexyl-Meldrum's acid (184.0 mg, 1.0 mmol, 1.0 equiv), *N*,*N*-diisopropylethylamine (522 μ L, 3.0 mmol, 3.0 equiv), and CH₃CN (1.0 mL). After the completion of the reaction, the workup method C was used to isolate **2b** in 95% yield as a sticky white solid (502.7 mg, 0.95 mmol). ¹⁹F NMR (376 MHz, CDCl₃): δ -56.0 (t, *J* = 21.6, 6.8 Hz, 3F), -135.2 to -135.4 (m, 2F), -144.7 to -145.1 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 3.5–3.3 (m, 2H), 3.0–2.9 (m, 2H), 2.0–1.8 (m, 4H), 1.5 (p, *J* = 5.9 Hz, 4H), 1.4–1.1 (m, 17H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 164.4, 146.8–143.7 (m), 143.7 (dd, *J* = 255.7, 17.5 Hz), 122.6 (d, *J* = 19.5 Hz), 121.2 (d, *J* = 234.0 Hz), 105.0 (qt, *J* = 33.8, 12.8 Hz), 102.7, 66.4, 54.1, 42.4, 34.5, 24.8, 22.4, 18.1, 16.7, 12.1. mp 100–101 °C. FT-IR (neat): cm⁻¹ 2936, 2706, 1593, 1286. HRMS (ESI) calcd, C₁₆H₁₀F₇O₄⁻ 399.0473; observed, 399.0448.

Note: Complete removal of DCM was not accomplished due to the extreme stickiness of the product. The yield was calculated taking the DCM into account.

Synthesis of *N*-Ethyl-*N*-isopropylpropan-2-aminium 4-Oxo-3-(2,3,5,6-tetrafluoro-4-nitrophenyl)-1,5-dioxaspiro[5.5]undec-2-en-2-olate (2c).

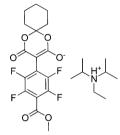




The Journal of Organic Chemistry

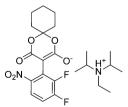
a brown sticky solid (253 mg, 0.5 mmol). The isolated product contained para and ortho substituted compounds in 7.1:1 ratio as determined by the ¹⁹F NMR of the isolated material. ¹⁹F NMR of the mixture (376 MHz, CDCl₃): δ -132.2 (minor, dd, J = 23.6, 9.5 Hz, 1F), -133.3 to -133.5 (major, m, 2F), -149.9 (minor, ddd, J = 21.7, 9.3, 4.1 Hz, 1F), -150.5 (major, td, J = 15.4, 14.7, 10.0 Hz, 2F), -152.1 to -152.4 (minor, m, 1F), -159.1 (minor, t, J = 21.2 Hz, 1F). ¹H NMR (400 MHz, CDCl₃): δ 3.58 (h, I = 10.2, 6.6, 3.7 Hz, 2H), 3.10-2.99 (m, 2H), 2.08-2.00 (m, 4H), 1.66 (p, J = 6.7 Hz, 4H), 1.52-1.34 (m, 17H). ¹³C{¹H} NMR of the mixture (101 MHz, $CDCl_{2}$: δ 164.6, 164.2, 146.5 (d, I = 6.3 Hz), 144.9 (ddd, I = 243.9, 10.5, 7.0 Hz), 142.7–139.6 (m), 140.2 (dd, J = 258.3, 18.8 Hz), 135.7–135.3 (m), 126.8 (t, J = 12.8 Hz), 123.4 (t, J = 18.5 Hz), 118.7 (dd, I = 20.4, 4.2 Hz), 102.8, 67.4, 66.6, 54.1, 53.5, 42.4, 35.1, 34.6,34.0, 24.8, 22.4, 18.2, 16.8, 12.2. mp 85-86 °C. FT-IR (neat): cm⁻¹ 2947, 1594, 1479, 1345. HRMS (ESI) calcd, C₁₅H₁₀F₄NO₆⁻ 376.0450; observed, 376.0424.

Synthesis of N-Ethyl-N-isopropylpropan-2-aminium 4-Oxo-3-(2,3,5,6-tetrafluoro-4-(methoxycarbonyl)phenyl)-1,5dioxaspiro[5.5]undec-2-en-2-olate (2d).



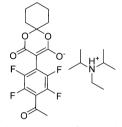
The general procedure A was followed using methyl-2,3,4,5,6pentafluorobenzoate (124.3 μ L, 0.55 mmol, 1.1 equiv), cyclohexyl-Meldrum's acid, MA (92.0 mg, 0.5 mmol, 1.0 equiv), *N*,*N*diisopropylethylamine (260 μ L, 1.5 mmol, 3.0 equiv), and CH₃CN (0.5 mL). After the completion of the reaction, the workup method C was used to isolate **2d** in 80% yield as colorless crystals (207.6 mg, 0.4 mmol). ¹⁹F NMR (376 MHz, CDCl₃): δ –136.2 to –136.3 (m, 2F), –142.8 to –143.0 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 3.93 (s, 3H), 3.58 (hept, *J* = 6.5 Hz, 2H), 3.09–2.99 (m, 2H), 2.09–1.98 (m, 4H), 1.71–1.56 (m, 4H), 1.45–1.29 (m, 17H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 164.5, 161.1, 146.7–143.7 (m), 146.1–143.0 (m), 121.6 (t, *J* = 18.7 Hz), 108.0 (t, *J* = 15.4 Hz), 102.5, 66.7, 54.0, 52.7, 42.3, 34.6, 24.9, 22.5, 18.2, 16.9, 12.1. mp 119–121 °C. FT-IR (neat): cm⁻¹ 2991, 2953, 1744, 1595, 1300. HRMS (ESI) calcd, C₁₇H₁₃F₄O₆⁻ 389.0654; observed, 389.0629.

Synthesis of *N*-Ethyl-*N*-isopropylpropan-2-aminium 3-(2,3-Difluoro-4-nitrophenyl)-4-oxo-1,5-dioxaspiro[5.5]undec-2-en-2-olate (2e).



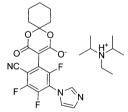
The general procedure A was followed using 1,2,3-trifluoro-4nitrobenzene (63.2 μ L, 0.55 mmol, 1.1 equiv), cyclohexyl-Meldrum's acid, MA (92.0 mg, 0.5 mmol, 1.0 equiv), *N*,*N*-diisopropylethylamine (260 μ L, 1.5 mmol, 3.0 equiv), and CH₃CN (0.5 mL). After the completion of the reaction, the workup method C was used to isolate **2e** in 98% yield as orange crystals (230.0 mg, 0.49 mmol). The isolated product contained *ortho* and *para* substituted compounds in 4.1:1 ratio as determined by the ¹⁹F NMR of the isolated material. ¹⁹F NMR of the mixture (376 MHz, CDCl₃): δ –129.5 (major, d, *J* = 22.5 Hz, 1F), –130.3 (minor, d, *J* = 27.1 Hz, 1F), –130.6 (major, d, *J* = 8.4 Hz, 1F), –144.6 (minor, d, *J* = 15.1 Hz, 1F). ¹H NMR (400 MHz, CDCl₃): δ 7.65–7.63 (minor, m, 2H), 7.57 (major, ddd, *J* = 9.1, 4.8, 1.8 Hz, 1H), 7.05–6.92 (major, m, 1H), 3.66–3.48 (major, m, 2H), 3.02 (major, q, *J* = 7.2 Hz, 2H), 2.16–1.95 (major, m, 4H), 1.72–1.57 (major, m, 4H), 1.50–1.27 (major, m, 17H). $^{13}C{^{1}H}$ NMR of the mixture (101 MHz, CDCl₃): δ 164.64, 164.55, 152.0 (dd, J = 407.5, 13.2 Hz), 152.0–147.2 (m), 151.8–147.1 (m), 151.7–147.2 (m), 149.5–146.7 (m), 149.4–149.1 (m), 146.9–146.8 (m), 146.2, 146.8–143.8 (m), 136.6 (d, J = 11.3 Hz), 132.8 (d, J = 4.0 Hz), 125.9, 124.3 (d, J = 15.8 Hz), 119.9 (dd, J = 8.3, 3.7 Hz), 118.6, 112.8 (d, J = 19.4 Hz), 102.4, 101.8, 73.7, 69.7, 53.9, 42.2, 35.3, 34.7, 34.0, 24.9, 22.5, 22.4, 22.4, 18.2, 16.9, 12.1. mp 130–133 °C. FT-IR (neat): cm⁻¹ 2991, 1678, 1592, 1213. HRMS (ESI) calcd, $C_{15}H_{12}F_2NO_6^{-3}$ 340.0638; observed, 340.0619.

Synthesis of *N*-Ethyl-*N*-isopropylpropan-2-aminium 3-(4-Acetyl-2,3,5,6-tetrafluorophenyl)-4-oxo-1,5-dioxaspiro[5.5]-undec-2-en-2-olate (2f).



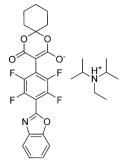
The general procedure A was followed using 2',3',4',5',6'-pentafluoroacetophenone (142.3 μ L, 1.0 mmol, 2.0 equiv), cyclohexyl-Meldrum's acid, MA (92.0 mg, 0.5 mmol, 1.0 equiv), *N*,*N*-diisopropylethylamine (260 μ L, 1.5 mmol, 3.0 equiv), and CH₃CN (0.5 mL). After the completion of the reaction, the workup method C was used to isolate **2f** in 95% yield as a yellow solid (239 mg, 0.48 mmol). ¹⁹F NMR (376 MHz, CDCl₃): δ –136.1 to –136.4 (m, 2F), –144.8 to –145.0 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 3.56 (hept, *J* = 6.6 Hz, 2H), 3.01 (q, *J* = 7.4 Hz, 2H), 2.07–2.01 (m, 4H), 2.00 (s, 3H), 1.66 (p, *J* = 6.4 Hz, 4H), 1.51–1.19 (m, 17H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 193.0, 164.5, 146.7–143.7 (m), 143.9 (dddd, *J* = 251.3, 16.3, 7.0, 4.0 Hz), 121.2 (t, *J* = 18.7 Hz), 115.4 (t, *J* = 16.1 Hz), 102.6, 66.8, 54.1, 42.4, 34.6, 32.4, 24.9, 22.5, 18.2, 16.8, 12.2. mp 58–60 °C. FT-IR (neat): cm⁻¹ 2943, 1599, 1314. HRMS (ESI) calcd, C₁₇H₁₃F₄O₅⁻ 373.0705; observed, 373.0677.

Synthesis of *N*-Ethyl-*N*-isopropylpropan-2-aminium 3-(2-Cyano-3,4,6-trifluoro-5-(1*H*-imidazol-1-yl)phenyl)-4-oxo-1,5-dioxaspiro[5.5]undec-2-en-2-olate (2g).



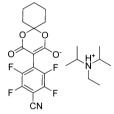
The general procedure A was followed using 2,3,5,6-tetrafluoro-4-(1Himidazol-1-yl)benzonitrile (132.5 mg, 0.55 mmol, 1.1 equiv), cyclohexyl-Meldrum's acid, MA (92.0 mg, 0.5 mmol, 1.0 equiv), N,Ndiisopropylethylamine (260 µL, 1.5 mmol, 3.0 equiv), and CH₃CN (0.5 mL). After the completion of the reaction, the workup method C was used to isolate 2g in 88% yield as a brown solid (236 mg, 0.44 mmol). ¹⁹F NMR (376 MHz, CDCl₃): δ –118.2 (d, J = 13.2 Hz, 1F), -134.9 (dd, J = 21.2, 13.2 Hz, 1F), -146.5 (d, J = 21.2 Hz, 1F). ¹H NMR (400 MHz, CDCl₃): δ 7.76 (d, J = 2.0 Hz, 1H), 7.21 (d, J = 1.5 Hz, 1H), 7.19 (s, 1H), 3.62 (hept, J = 6.7 Hz, 2H), 3.07 (q, J = 7.4 Hz, 2H), 2.15-1.90 (m, 4H), 1.75-1.11 (m, 21H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 164.5, 151.2, 148.7, 149.8–142.5 (m), 147.1–140.0 (m), 129.3, 127.6 (dd, J = 20.9, 3.8 Hz), 120.1, 119.7 (dd, J = 19.9, 11.4 Hz), 112.2 (t, J = 3.6 Hz), 105.0 (dd, J = 10.5, 5.5 Hz), 102.7, 70.4, 53.9, 42.2, 34.6 (d, J = 14.7 Hz), 24.7, 22.4, 18.3, 17.0, 12.1. mp 80-82 °C. FT-IR (neat): cm⁻¹ 2938, 1681, 1589, 1482. HRMS (ESI) calcd, C₁₉H₁₃F₃N₃O₄⁻ 404.0864; observed, 404.0852.

Synthesis of *N*-Ethyl-*N*-isopropylpropan-2-aminium 3-(4-(2,7*a*-Dihydrobenzo[*d*]oxazol-2-yl)-2,3,5,6-tetrafluorophenyl)-4-oxo-1,5-dioxaspiro[5.5]undec-2-en-2-olate (2h).



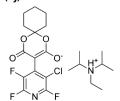
The general procedure A was followed using 2-(perfluorophenyl)benzo[d]oxazole (156.8 mg, 0.55 mmol, 1.1 equiv), cyclohexyl-Meldrum's acid, MA (92.0 mg, 0.5 mmol, 1.0 equiv), N,Ndiisopropylethylamine (260 μ L, 1.5 mmol, 3.0 equiv), and CH₃CN (0.5 mL). After the completion of the reaction, the workup method C was used to isolate 2h in 82% yield as a yellow oil (237 mg, 0.41 mmol). ¹⁹F NMR (376 MHz, CDCl₃): δ –135.9 to –136.0 (m, 2F), -141.6 to -141.7 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 7.89–7.61 (m, 2H), 7.46–7.35 (m, 2H), 3.61 (hept, J = 6.8, 6.3 Hz, 2H), 3.05 (q, I = 7.3 Hz, 2H), 2.11–2.04 (m, 4H), 1.67 (m, 4H), 1.47–1.33 (m, 17H). ${}^{13}C{}^{1}H}$ NMR (101 MHz, CDCl₃): δ 164.4, 154.2, 150.1, 145.5 (dddd, J = 244.3, 11.3, 6.2, 3.3 Hz), 144.7 (ddt, J = 255.5, 16.7, 4.9 Hz), 141.0, 125.8, 124.7, 121.3 (t, J = 18.7 Hz), 120.2, 110.6, 103.6 (t, J = 13.7 Hz), 102.5, 66.9, 54.0, 42.3, 34.5, 24.8, 22.4, 18.1, 16.7, 12.1. FT-IR: cm⁻¹ 2942, 1686, 1597, 1488, 1304. HRMS (ESI) calcd, C₂₂H₁₆F₄NO₅⁻ 450.0970; observed, 450.0950.

Synthesis of *N*-Ethyl-*N*-isopropylpropan-2-aminium 3-(4-Cyano-2,3,5,6-tetrafluorophenyl)-4-oxo-1,5-dioxaspiro[5.5]-undec-2-en-2-olate (2i).



This reaction was carried out in a borosilicate test tube. The general procedure A was followed using 2,3,4,5,6-pentafluorobenzonitrile (1.0 g, 5.42 mmol, 2.0 equiv), cyclohexyl-Meldrum's acid, MA (0.5 g, 2.71 mmol, 1.0 equiv), *N*,*N*-diisopropylethylamine (1.41 mL, 8.1 mmol, 3.0 equiv), and CH₃CN (27 mL). After the completion of the reaction, the workup method C was used to isolate **2i** in 98% yield as a yellow solid (1.3 g, 2.66 mmol). ¹⁹F NMR (376 MHz, CDCl₃): δ –133.3 to –133.6 (m, 2F), –136.4 to –136.7 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 3.60 (hept, *J* = 10.0, 6.5, 3.2 Hz, 2H), 3.05 (qd, *J* = 7.4, 3.6 Hz, 2H), 2.06–2.00 (m, 4H), 1.66 (p, *J* = 6.7 Hz, 4H), 1.44–1.34 (m, 17H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 163.5, 147.7–144.5 (m), 145.8–142.9 (m), 125.2 (t, *J* = 18.2 Hz), 108.2, 102.2, 88.3, 66.7, 53.7, 42.0, 34.1, 24.3, 21.9, 17.6, 16.2, 11.7. mp 72–74 °C. FT-IR: cm⁻¹ 2948, 1594, 1481, 1291. HRMS (ESI) calcd, C₁₆H₁₀F₄NO₄⁻ 356.0551; observed, 356.0521.

Synthesis of *N*-Ethyl-*N*-isopropylpropan-2-aminium 3-(3-Chloro-2,5,6-trifluoropyridin-4-yl)-4-oxo-1,5-dioxaspiro[5.5]-undec-2-en-2-olate (2j).



The general procedure A was followed using 3-chloro-2,4,5,6-tetrafluoropyridine (22.6 μ L, 0.2 mmol, 2.0 equiv), cyclohexyl-

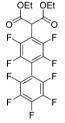
Meldrum's acid, MA (18.42 mg, 0.1 mmol, 1.0 equiv), *N*,*N*-diisopropylethylamine (52.0 μ L, 0.3 mmol, 3.0 equiv), and CH₃CN (0.5 mL). After the completion of the reaction, the workup method B was used to isolate **2j** in 94% yield as a colorless oil (45.0 mg, 0.09 mmol). ¹⁹F NMR (376 MHz, CDCl₃): δ –76.6 to –77.0 (m, 1F), –93.4 to –93.6 (m, 1F), –138.1 (t, *J* = 24.6 Hz, 1F). ¹H NMR (400 MHz, CDCl₃): δ 3.64–3.50 (m, 2H), 3.04 (qd, *J* = 7.4, 4.0 Hz, 2H), 2.08–1.98 (m, 4H), 1.65 (p, *J* = 6.6 Hz, 4H), 1.50–1.32 (m, 17H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 164.0, 150.8 (ddd, *J* = 239.2, 13.7, 2.5 Hz), 146.9 (ddd, *J* = 243.2, 18.9, 15.2 Hz), 141.5 (ddd, *J* = 252.6, 24.4, 5.9 Hz), 115.4 (dd, *J* = 31.1, 5.9 Hz), 103.0, 71.2, 54.1, 42.4, 35.7, 33.8, 25.0, 22.6 (d, *J* = 22.6 Hz), 18.4, 17.1, 12.2. FT-IR: cm⁻¹ 2942,2 1599, 1434, 1216. HRMS (ESI) calcd, C₁₄H₁₀ClF₃NO₄⁻ 348.0256; observed, 348.0248.

Synthesis of Diethyl 2-(Perfluorophenyl)malonate (3k).



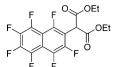
The general procedure D was followed using hexafluorobenzene (138 μ L, 1.2 mmol, 1.2 equiv), diethyl malonate (153 μ L, 1.0 mmol, 1.0 equiv), K₂CO₃ (414 mg, 3.0 mmol, 3.0 equiv), and DMF (0.6 mL). The crude material was purified by flash chromatography using hexane:ethyl acetate (0% EtOAc for 0-4 cv, 0-10% EtOAc for 4-9 cv, 10% EtOAc for 9–11 cv, 10–100% EtOAc for 11–16 cv, then held at 100% EtOAc 16-18 cv), on 4 g silica column to afford 3a in 92% yield (300 mg, 0.92 mmol) as a colorless liquid. ¹⁹F NMR (376 MHz, $CDCl_3$): $\delta - 139.9$ to - 140.1 (m, 2F), - 153.4 (ddd, J = 21.0, 19.1, 1.7Hz, 1F), -161.6 to -161.8 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 4.93 (s, 1H), 4.27 (qd, J = 7.1, 2.0 Hz, 4H), 1.29 (t, J = 7.1 Hz, 6H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 165.3, 145.4 (dtt, *J* = 249.8, 7.6, 3.7 Hz), 141.2 (dtd, J = 254.9, 13.4, 6.8 Hz), 139.1-136.0 (m), 108.4 (td, J = 17.0, 3.9 Hz), 62.5, 46.9, 13.6. FT-IR: cm⁻¹ 2987, 1751, 1508, 1307, 1224. HRMS (ESI) calcd, C13H11F5O4 326.0577; observed, 326.0562.

Synthesis of Diethyl 2-(Perfluoro-[1,1'-biphenyl]-4-yl)malonate (3l).



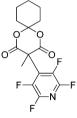
The general procedure D was followed using decafluorobiphenyl (80.2 mg, 0.24 mmol, 1.2 equiv), diethyl malonate (30.5 μ L, 0.2 mmol, 1.0 equiv), K₂CO₃ (83 mg, 0.6 mmol, 3.0 equiv), and DMF (0.6 mL). The crude material was purified by flash chromatography using hexane:ethyl acetate (0% EtOAc for 0-5 cv, 0%-9% EtOAc for 5-10 cv, 9% EtOAc for 10-15 cv, 9%-17% EtOAc for 15-18 cv, 17% EtOAc for 18-22 cv, 17%-100% EtOAc for 22-25 cv, then held at 100% EtOAc 25-27 cv), on 4 g silica column to afford 3l in 73% yield (69.0 mg, 0.15 mmol) as white crystals. ¹⁹F NMR (376 MHz, CDCl₃): δ -137.0 (dddd, J = 17.4, 11.8, 8.7, 5.8 Hz, 2F), -137.9 to -138.1 (m, 2F), -139.1 to -139.3 (m, 2F), -150.0 (tt, J = 20.8, 3.1 Hz, 1F), -160.3 to -160.5 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 5.05 (s, 1H), 4.32 (qq, J = 7.0, 3.6 Hz, 4H), 1.33 (t, J = 7.1 Hz, 6H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 165.3, 146.7–143.7 (m), 146.0–143.0 (m), 144.0 (ddt, J = 252.0, 14.9, 4.4 Hz), 139.6-138.7 (m), 141.5-136.2 (m), 115.4 (t, J = 16.6 Hz), 107.0–101.7 (m), 67.1, 62.8, 47.6, 13.9. mp 83-84 °C. FT-IR: cm⁻¹ 2989, 2926, 1739, 1477. HRMS (ESI) calcd, C₁₉H₁₁F₉O₄ 474.0514; observed, 474.0523.

Synthesis of Diethyl 2-(Perfluoronaphthalen-2-yl)malonate (3m).



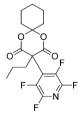
The general procedure D was followed using octafluoronaphthalene (65.3 mg, 0.24 mmol, 1.2 equiv), diethyl malonate (30.5 *µ*L, 0.2 mmol, 1.0 equiv), K₂CO₃ (83 mg, 0.6 mmol, 3.0 equiv), and DMF (0.6 mL). The crude material was purified by flash chromatography using hexane:ethyl acetate (0% EtOAc for 0-4 cv, 0-10% EtOAc for 4-9 cv, 10% EtOAc for 9-11 cv, 10-100% EtOAc for 11-16 cv, then held at 100% EtOAc 16-18 cv), on 4 g silica column to afford 3m in 75% yield (69.0 mg, 0.15 mmol) as a colorless liquid. ¹⁹F NMR (376 MHz, CDCl₃): δ -118.8 to -119.3 (m, 1F), -135.2 to -135.5 (m, 1F), -143.3 to -143.8 (m, 1F), -145.6 to -146.1 (m, 1F), -148.1 to -148.6 (m,1F), -152.6 (q, J = 20.8 Hz, 1F), -155.2 (d, J = 19.4 Hz,1F). ¹H NMR (400 MHz, CDCl₃): δ 5.13 (s, 1H), 4.30 (q, J = 7.3, 6.5 Hz, 4H), 1.30 (t, J = 7.1 Hz, 6H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 165.6, 152.4-148.9 (m), 147.9-142.6 (m), 145.2-142.2 (m), 141.5-141.0 (m), 140.1-140.0 (m), 140.4-139.7 (m), 139.0-137.0 (m), 112.3–111.7 (m), 152.2–149.2 (m), 111.7–107.4 (m), 62.7, 47.5, 13.9. FT-IR: cm^{-1} 2987, 2942, 1745, 1655, 1497, 1441, 1409, 1181. HRMS (ESI) calcd, C₁₇H₁₁F₇O₄ 412.0546; observed, 474.0541. Synthesis of 3-Methyl-3-(perfluoropyridin-4-yl)-1,5-

dioxaspiro[5.5]undecane-2,4-dione (5aa).



The general procedure E was followed using pentafluoropyridine (54.9 µL, 0.5 mmol, 2.0 equiv), 3-methyl-1,5-dioxaspiro[5.5]undecane-2,4dione (49.25 mg, 0.25 mmol, 1.0 equiv), N,N-diisopropylethylamine (130.4 µL, 0.75 mmol, 3.0 equiv), and CH₃CN (0.5 mL). After the completion of the reaction, the workup method B was used to isolate 5aa in 96% yield as a white solid (83. $\overline{6}$ mg, 0.24 mmol). The isolated product contained a mixture of the para and ortho isomers in 25:1 ratio as determined by the ¹⁹F NMR of the isolated material. ¹⁹F NMR $(376 \text{ MHz}, \text{CDCl}_3)$: $\delta - 89.1 - 89.3 \text{ (m, 2F)}, -138.9 \text{ to } -139.2 \text{ (m, 2F)}, -138.9 \text{ (m, 2F)}, -138.9 \text{ (m, 2F)}, -138.9 \text{ (m, 2F)}, -1$ 2F). ¹H NMR (400 MHz, CDCl₃): δ 2.25 (t, J = 3.0 Hz, 3H), 2.05 (m, 4H), 1.87-1.74 (m, 4H), 1.60-1.50 (m, 2H). ¹³C{¹H} NMR (101 MHz, $CDCl_3$): δ 165.6, 145.5–142.5 (m), 142.6–139.5 (m), 129.2 (tt, J = 11.7, 2.2 Hz), 109.0, 53.7, 37.5, 24.0 (t, J = 6.5 Hz), 23.8, 22.2 (d, J = 8.1 Hz). mp 100–101 °C. FT-IR: cm⁻¹ 2998, 2947, 1741, 1453, 1261. HRMS (ESI) calcd, $C_{15}H_{13}F_4NO_4$ 347.0781; observed, 347.0772.

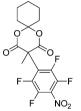
Synthesis of 3-(Perfluoropyridin-4-yl)-3-propyl-1,5dioxaspiro[5.5]undecane-2,4-dione (5ab).



The general procedure E was followed using pentafluoropyridine (54.9 μ L, 0.5 mmol, 2.0 equiv), 3-propyl-1,5-dioxaspiro[5.5]undecane-2,4-dione (56.57 mg, 0.25 mmol, 1.0 equiv), *N*,*N*-diisopropylethylamine (130.4 μ L, 0.75 mmol, 3.0 equiv), and CH₃CN (0.5 mL). After the completion of the reaction, the workup method B was used to isolate **Sab** in >99% yield as a colorless liquid (94 mg, 0.25 mmol). The isolated product contained a mixture of the *para* and *ortho* isomers in

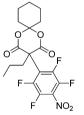
25:1 ratio as determined by the ¹⁹F NMR of the isolated material. ¹⁹F NMR (376 MHz, CDCl₃): δ –89.0 to –89.5 (m, 2F), –137.5 (dd, *J* = 35.0, 14.9 Hz, 2F). ¹H NMR (400 MHz, CDCl₃): δ 2.60–2.49 (m, 2H), 2.02 (m, 4H), 1.85–1.70 (m, 4H), 1.58–1.41 (m, 4H), 1.01 (t, *J* = 7.3 Hz, 3H). ¹³C{¹H} NMR of the mixture (101 MHz, CDCl₃): δ 161.8, 160.7, 141.8–138.8 (m), 138.7–135.5 (m), 124.6 (t, *J* = 11.5 Hz), 105.0, 101.7, 35.2 (t, *J* = 5.3 Hz), 34.4 (t, *J* = 2.4 Hz), 33.4, 33.0, 32.1, 20.1, 19.9, 18.6, 18.5, 18.1, 17.8, 16.0, 15.5, 10.0, 9.7. FT-IR: cm⁻¹ 2944, 2874, 1746, 1469, 1259. HRMS (ESI) calcd, C₁₇H₁₇F₄NO₄ 375.1094; observed, 375.1063.

Synthesis of 3-Methyl-3-(2,3,5,6-tetrafluoro-4-nitrophenyl)-1,5-dioxaspiro[5.5]undecane-2,4-dione (5ca).



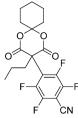
The general procedure E was followed using pentafluoronitrobenzene (33.3 µL, 0.275 mmol, 1.1 equiv), 3-methyl-1,5-dioxaspiro[5.5]undecane-2,4-dione (49.32 mg, 0.25 mmol, 1.0 equiv), N,Ndiisopropylethylamine (130.4 µL, 0.75 mmol, 3.0 equiv), and CH_2CN (0.5 mL). After the completion of the reaction, the workup method B was followed. The crude material was purified by flash chromatography using hexane:ethyl acetate (0% EtOAc for 0-2 cv, 0-10% EtOAc for 2-6 cv, 10% EtOAc for 6-11 cv, 10-100% EtOAc for 11-16 cv, then held at 100% EtOAc 16-18 cv), on 4 g silica column to afford 5ca in 87% yield (85.0 mg, 0.21 mmol) as a colorless oil. ¹⁹F NMR (376 MHz, CDCl₃): δ -134.3 (m, 2F), -145.3 to -145.5 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 2.2 (t, J = 3.1 Hz, 3H), 2.1–2.0 (m, 4H), 1.9–1.8 (m, 4H), 1.6–1.5 (m, 2H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 166.0, 146.1 (ddt, J = 254.0, 12.9, 5.6 Hz), 140.6 (ddd, J = 263.2, 17.5, 3.7 Hz), 120.5 (t, J = 13.1 Hz), 109.1, 51.2, 38.1, 37.6, 24.3 (t, J = 7.0 Hz), 24.0, 22.4 (d, J = 8.6 Hz). FT-IR: cm⁻¹ 2947, 1743, 1554, 1305. HRMS (ESI) calcd, C₁₆H₁₃F₄NO₆ 391.0679; observed, 391.0665.

Synthesis of 3-Propyl-3-(2,3,5,6-tetrafluoro-4-nitrophenyl)-1,5-dioxaspiro[5.5]undecane-2,4-dione (5cb).



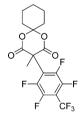
The general procedure E was followed using pentafluoronitrobenzene (33.3 µL, 0.275 mmol, 1.1 equiv), 3-propyl-1,5-dioxaspiro[5.5]undecane-2,4-dione (49.32 mg, 0.25 mmol, 1.0 equiv), N,Ndiisopropylethylamine (130.4 μ L, 0.75 mmol, 3.0 equiv), and CH₃CN (0.5 mL). After the completion of the reaction, the workup method B was followed. The crude material was purified by flash chromatography using hexane:ethyl acetate (0% EtOAc for 0-3 cv, 0-10% EtOAc for 3-7 cv, 10% EtOAc for 7-11 cv, 10-30% EtOAc for 11-15 cv, 30-100% EtOAc for 15-16, then held at 100% EtOAc 16-18 cv), on 4 g silica column to afford 5cb in 81% yield (84.2 mg, 0.20 mmol) as yellow crystals. ¹⁹F NMR (376 MHz, \dot{CDCl}_3): δ –130.9 to -131.7 (m, 2F), -133.3 to -133.7 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 2.58–2.46 (m, 2H), 2.11–1.92 (m, 4H), 1.86–1.71 (m, 4H), 1.59–1.43 (m, 4H), 1.03 (t, J = 7.3 Hz, 3H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 165.0, 146.1 (ddt, J = 253.7, 13.0, 5.8 Hz), 142.5– 139.1 (m), 120.0 (t, J = 13.0 Hz), 109.1, 39.3 (t, J = 5.7 Hz), 38.4 (t, J = 2.3 Hz), 37.5, 24.0, 22.6, 22.2, 19.6, 13.7. mp 99-100 °C. FT-IR: cm^{-1} 2953, 1737, 1253. HRMS (ESI) calcd, $C_{18}H_{17}F_4NO_6H$ 420.1070; observed, 420.1051.

Synthesis of 4-(2,4-Dioxo-3-propyl-1,5-dioxaspiro[5.5]undecan-3-yl)-2,3,5,6-tetrafluorobenzonitrile (5ib).



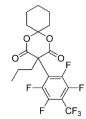
The general procedure E was followed using pentafluorobenzonitrile (34.7 μL, 0.275 mmol, 1.1 equiv), 3-propyl-1,5-dioxaspiro[5.5]undecane-2,4-dione (49.32 mg, 0.25 mmol, 1.0 equiv), N,Ndiisopropylethylamine (130.4 µL, 0.75 mmol, 3.0 equiv), and CH₃CN (0.5 mL). After the completion of the reaction, the workup method B was followed. The crude material was purified by flash chromatography using hexane:ethyl acetate (0% EtOAc for 0-5 cv, 0-10% EtOAc for 5-25 cv, 10% EtOAc for 25-40 cv, 10-40% EtOAc for 40-60 cv, 40-100% EtOAc for 60-65, then held at 100% EtOAc 65-67 cv), on 4 g silica column to afford 5ib in 85% yield (85.0 mg, 0.21 mmol) as a colorless oil. ¹⁹F NMR (376 MHz, CDCl₃): δ –126.7 to -136.0 (m, 2F), -142.2 to -151.4 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 2.57–2.47 (m, 2H), 2.12–1.95 (m, 4H), 1.87–1.73 (m, 4H), 1.60–1.43 (m, 4H), 1.03 (t, J = 7.3 Hz, 3H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 164.8, 147.5 (ddt, J = 262.8, 17.8, 3.8 Hz), 145.9 (ddt, J = 252.3, 12.6, 5.3 Hz), 122.1 (d, J = 12.9 Hz), 108.9, 106.7 (t, J = 3.7 Hz), 39.1 (t, J = 5.7 Hz), 38.3, 37.4, 23.9, 22.5, 22.1, 19.5, 13.6. FT-IR: cm⁻¹ 2944, 1744, 1491. HRMS (ESI) calcd, C₁₉H₁₇F₄NO₄ 399.1094; observed, 399.1081.

Synthesis of 3-Methyl-3-(2,3,5,6-tetrafluoro-4-(trifluoromethyl)phenyl)-1,5-dioxaspiro[5.5]undecane-2,4dione (5ba).



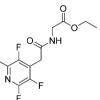
The general procedure E was followed using octafluorotoluene (70.8 µL, 0.50 mmol, 2.0 equiv), 3-methyl-1,5-dioxaspiro[5.5]undecane-2,4dione (49.32 mg, 0.25 mmol, 1.0 equiv), N,N-diisopropylethylamine (130.4 µL, 0.75 mmol, 3.0 equiv), and CH₃CN (0.5 mL). After the completion of the reaction, the workup method B was followed. The crude material was purified by flash chromatography using hexane:ethyl acetate (0% EtOAc for 0-6 cv, 0-10% EtOAc for 6-23 cv, 10% EtOAc for 23-35 cv, 10-100% EtOAc for 35-45 cv, then held at 100% EtOAc 45-47 cv), on 4 g silica column to afford 5ba in 80% yield (83.0 mg, 0.20 mmol) as a white solid. ¹⁹F NMR (376 MHz, $CDCl_3$): δ -56.6 (t, J = 21.4 Hz, 3F), -136.0 to -136.2 (m, 2F), -139.2 to -139.5 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 2.24 (t, J = 3.0 Hz, 3H), 2.12-2.00 (m, 4H), 1.86-1.75 (m, 4H), 1.58-1.50 (m, 2H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 166.1, 146.0 (ddt, J = 251.2, 14.8, 5.2 Hz), 144.4 (dd, J = 261.6, 17.5 Hz), 121.8, 120.4 (t, J = 13.0 Hz), 119.0, 108.8, 37.9, 37.6, 24.2 (t, J = 6.9 Hz), 23.9, 22.2 (d, J = 8.2 Hz). mp 100–102 °C. FT-IR: cm⁻¹ 2944, 2856, 1736, 1142. HRMS (ESI) calcd, C₁₇H₁₃F₇O₄H 415.0780; observed, 414.0792.

Synthesis of 3-Propyl-3-(2,3,5,6-tetrafluoro-4-(trifluoromethyl)phenyl)-1,5-dioxaspiro[5.5]undecane-2,4dione (5bb).



The general procedure E was followed using octafluorotoluene (70.8 µL, 0.50 mmol, 2.0 equiv), 3-propyl-1,5-dioxaspiro[5.5]undecane-2,4dione (49.32 mg, 0.25 mmol, 1.0 equiv), N,N-diisopropylethylamine (130.4 μ L, 0.75 mmol, 3.0 equiv), and CH₃CN (0.5 mL). After the completion of the reaction, the workup method B was followed. The crude material was purified by flash chromatography using hexane:ethyl acetate (0% EtOAc for 0-2 cv, 0-10% EtOAc for 2-6 cv, 10% EtOAc for 6-8 cv, 10-15% EtOAc for 8-10 cv, 15-100% EtOAc for 10-12 cv, then held at 100% EtOAc 12-14 cv), on 4 g silica column to afford 5bb in 79% yield (88.3 mg, 0.19 mmol) as a colorless oil. ¹⁹F NMR (376 MHz, CDCl₃): δ -56.7 (t, J = 21.3 Hz, 3F), -134.4 to -134.9 (m, 2F), -139.0 to -139.6 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 2.53-2.41 (m, 2H), 2.05-1.89 (m, 4H), 1.82-1.65 (m, 4H), 1.53–1.36 (m, 4H), 0.96 (t, J = 7.3 Hz, 3H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 165.3, 147.7-144.6 (m), 146.4-143.0 (m), 120.0 (t, J = 12.9 Hz), 111.1–110.2 (m), 108.9, 39.4 (t, J = 5.6Hz), 38.4 (t, J = 2.5 Hz), 37.6, 24.1, 22.6, 22.2, 19.6, 13.8. FT-IR: cm⁻¹ 2946, 2875, 1747, 1492, 13334. HRMS (ESI) calcd, C10H17F7O4H 443.1093; observed, 443.1095.

Synthesis of Ethyl 2-(2-(Perfluoropyridin-4-yl)acetamido)-acetate (6aa).



The general procedure E was followed using **2a** (46.2 mg, 0.1 mmol, 1.0 equiv) and ethyl 2-aminoacetate hydrochloride (41.9 mg, 0.3 mmol, 3.0 equiv) and dioxane (1.0 mL) to obtain **6aa** in 94% yield (55.0 mg, 0.09 mmol) as an off-white solid. ¹⁹F NMR (376 MHz, CDCl₃): δ –91.0 to –91.3 (m, 2F), –143.4 to –143.6 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 4.20 (q, *J* = 7.2 Hz, 2H), 4.03 (d, *J* = 5.1 Hz, 2H), 3.81 (s, 2H), 1.26 (t, *J* = 7.1 Hz, 3H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 169.7, 166.1, 144.8–141.7 (m), 142.4–141.8 (m), 128.0 (tt, *J* = 16.7, 2.9 Hz), 61.9, 41.8, 30.4, 14.1 mp 90–92 °C. FT-IR: cm⁻¹ 3316, 2997, 1741, 1459, 1225. HRMS (ESI) calcd, C₁₁H₁₀F₄N₂O₃ 294.0628; observed, 294.0642.

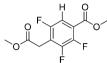
Synthesis of Methyl 2,3,5,6-Tetrafluoro-4-(2-methoxy-2-oxoethyl)benzoate (6dd).



For the monitoring purposes, this reaction was carried out in an NMR tube and monitored via ¹⁹F NMR. An NMR tube was charged with **2d** (100 mg, 0.2 mmol, 1.0 equiv), and HCl (36% w/v, 20.6 μ L, 0.24 mmol, 1.2 equiv) and methanol (0.5 mL) were added, and a sealed glass capillary containing C₆D₆ was placed in the NMR tube for locking purposes. The NMR tube then was placed in a sand bath and maintained at 80 °C. The reaction was periodically monitored by ¹⁹F NMR. After the completion of the reaction, methanol was removed in vacuo, and the resulting residue was dissolved in DCM (2 mL) and washed with deionized water (3 × 5 mL). The organic layer was dried with anhydrous MgSO₄, filtered, and concentrated in vacuo to obtain

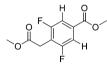
the title compound **6dd** in 95% yield (53.0 mg, 0.19 mmol) as a colorless liquid. ¹⁹F NMR (376 MHz, CDCl₃): δ –139.7 to –139.9 (m, 2F), –141.4 to –141.5 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 3.98 (s, 3H), 3.80 (t, J = 1.6 Hz, 2H), 3.75 (s, 3H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 168.2, 160.0, 145.1 (ddt, J = 248.6, 14.3, 5.1 Hz), 144.4 (ddt, J = 256.5, 15.4, 4.6 Hz), 116.5 (t, J = 18.2 Hz), 111.8 (t, J = 15.9 Hz), 53.2, 52.7, 28.2. FT-IR: cm⁻¹ 2960, 1742, 1485, 1310. HRMS (ESI) calcd, C₁₁H₈F₄O₄ 280.0359; observed, 280.0382.

Synthesis of Methyl 2,3,5-Trifluoro-4-(2-methoxy-2oxoethyl)benzoate (6de).



The general procedure F was followed using methyl 2,3,5,6tetrafluoro-4-(2-methoxy-2-oxoethyl)benzoate (6dd) (28.0 mg, 0.1 mmol, 1.0 equiv) and N,N-diisopropylethylamine (19.2 µL, 0.11 mmol, 1.1 equiv). After the completion of the reaction, the crude material was purified by flash chromatography using hexane:ethyl acetate (0% EtOAc for 0–1 cv, 0–11% EtOAc for 1–3 cv, 11% EtOAc for 3-4 cv, 11-22% EtOAc for 4-5 cv, 22-100% EtOAc for 5-8 cv, then held at 100% EtOAc 8-10 cv), on 4 g silica column to afford 6de in 94% yield (25.0 mg, 0.09 mmol) as a colorless oil. The isolated product also contained 2% starting material and 4% dihydrodefluorinated product. ¹⁹F NMR (376 MHz, CDCl₂): δ –118.7 to –118.8 (m, 1F), -135.0 (d, J = 20.3 Hz, 1F), -139.0 (ddd, J = 20.7, 16.1, 5.1 Hz, 1F). ¹H NMR (400 MHz, CDCl₃): δ 7.43-7.36 (m, 1H), 3.88 (s, 3H), 3.71 (s, 2H), 3.67 (s, 3H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 168.7, 163.0 (q, J = 3.4 Hz), 155.6 (ddd, J = 246.5, 6.5, 3.5 Hz), 149.9 (ddd, J = 251.4, 15.0, 8.1 Hz), 147.0 (ddd, J = 259.8, 14.5, 3.8 Hz), 119.3 (t, J = 8.9 Hz), 117.2 (dd, J = 22.2, 16.8 Hz), 112.3 (dd, J = 26.1, 3.8 Hz), 52.9, 52.6, 28.4–28.3 (m). FT-IR: cm⁻¹ 2958, 1742, 1468, 1265. HRMS (ESI) calcd, C11HoF3O4 262.0453; observed, 262.0469.

Synthesis of Methyl 3,5-Difluoro-4-(2-methoxy-2-oxoethyl)benzoate (6df).



The general procedure F was followed using methyl-2,3,5,6-tetrafluoro-4-(2-methoxy-2-oxoethyl)benzoate (**6dd**) (26.5 mg, 0.09 mmol, 1.0 equiv) and *N*,*N*-diisopropylethylamine (99.0 μ L, 0.57 mmol, 6.0 equiv). After the completion of the reaction, the crude material was purified by passing through a short silica plug to afford **6df** in 92% yield (23.2 mg, 0.08 mmol) as a colorless oil. ¹⁹F NMR (376 MHz, CDCl₃): δ –112.9 (d, *J* = 7.3 Hz, 2F). ¹H NMR (400 MHz, CDCl₃): δ 7.60–7.54 (m, 2H), 3.92 (s, 3H), 3.75 (s, 2H), 3.72 (s, 3H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 164.8 (t, *J* = 3.3 Hz), 161.1 (dd, *J* = 249.6, 7.9 Hz), 131.6 (t, *J* = 9.5 Hz), 115.5 (t, *J* = 20.4 Hz), 112.4 (d, *J* = 27.1 Hz), 112.4 (d, *J* = 12.5 Hz), 52.6, 52.5, 28.0. FT-IR: cm⁻¹ 2957, 1730, 1586, 1430. HRMS (ESI) calcd, C₁₁H₁₀F₂O₄ 244.0547; observed, 244.0529.

Synthesis of N-Methoxy-N-methyl-2-(perfluoropyridin-4-yl)acetamide (6ac).



The general procedure E was followed using **2a** (23.1 mg, 0.05 mmol, 1.0 equiv) and *N*,*O*-dimethylhydroxylamine hydrochloride (11.8 mg, 0.12 mmol, 2.4 equiv) and dioxane (0.5 mL) to obtain **6ac** in 90% yield (11.3 mg, 0.04 mmol) as a yellow liquid. ¹⁹F NMR (376 MHz, CDCl₃): δ –91.0 to –91.3 (m, 2F), –143.6 to –143.9 (m, 2F). ¹H NMR (400 MHz, CDCl₃): δ 3.98 (s, 2H), 3.81 (s, 3H), 3.25 (s, 3H). FT-IR: cm⁻¹ 2937, 1710, 1471, 1047. HRMS (ESI) calcd, C₉H₈F₄N₂O₂ 252.0522; observed, 252.0529.

Synthesis of 3-Methyl-1,5-dioxaspiro[5.5]undecane-2,4-dione (4a).



The general procedure G was followed using formic acid (186 μ L, 4.93 mmol), cyclohexyl-Meldrum's acid (1.0 g, 5.43 mmol), DMAP (60 mg, 0.49 mmol), *N*,*N*-diisopropylethylamine (1.8 mL, 10.6 mmol), DCC (1.4 g, 5.43 mmol), acetic acid (2.8 mL, 49.3 mmol), NaBH₄ (466 mg, 12.3 mmol), and MeCN (25 mL). The crude material was purified by flash chromatography using hexane:ethyl acetate (0% EtOAc for 0–5 cv, 0–12% EtOAc for 5–15 cv, 12% EtOAc for 15–25 cv, 12–50% EtOAc for 25–30 cv, 50–100% EtOAc for 30–37 cv, then held at 100% EtOAc 37–39 cv), on 40 g silica column to afford **4a** in 71% yield (971 mg, 3.5 mmol) as white crystals. ¹H NMR (400 MHz, CDCl₃): δ 3.61 (q, *J* = 7.1 Hz, 1H), 2.07–1.91 (m, 4H), 1.84–1.64 (m, 4H), 1.63–1.46 (m, 5H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 166.1, 105.6, 41.4, 36.9, 35.1, 24.0, 22.5, 21.8, 10.6, which matched with the literature.³¹

Synthesis of 3-Propyl-1,5-dioxaspiro[5.5]undecane-2,4-dione (4b).

The general procedure G was followed using propionic acid (369 μ L, 4.93 mmol), cyclohexyl-Meldrum's acid (1.0 g, 5.43 mmol), DMAP (60 mg, 0.49 mmol), *N*,*N*-diisopropylethylamine (1.8 mL, 10.6 mmol), DCC (1.4 g, 5.43 mmol), acetic acid (2.8 mL, 49.3 mmol), and NaBH₄ (466 mg, 12.3 mmol). The crude material was purified by flash chromatography using hexane:ethyl acetate (0% EtOAc for 0–3 cv, 0–9% EtOAc for 3–11 cv, 9% EtOAc for 11–30 cv, 9–35% EtOAc for 30–37 cv, 35% EtOAc for 37–45 cv, 35–70% EtOAc for 45–50 cv, 70% EtOAc for 50–55 cv, 70–100% EtOAc for 55–65 cv, then held at 100% EtOAc 65–67 cv), on 40 g silica column to afford **4b** in 92% yield (1.0 g, 4.5 mmol) as white crystals. ¹H NMR (400 MHz, CDCl₃): δ 3.51 (t, *J* = 5.1 Hz, 1H), 2.11–2.03 (m, 2H), 2.03–1.89 (m, 4H), 1.83–1.62 (m, 4H), 1.55–1.39 (m, 4H), 0.96 (t, *J* = 7.3 Hz, 3H). ¹³C{¹H} NMR (101 MHz, CDCl₃): δ 165.6, 105.5, 46.1, 36.8, 35.7, 28.6, 24.0, 22.5, 21.7, 19.7, 13.8, which matched with the literature.³¹

ASSOCIATED CONTENT

S Supporting Information

Spectra and additional experimental details. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

Oklahoma State University is gratefully acknowledged for startup funds.

REFERENCES

- (1) Meldrum, A. N. J. Chem. Soc., Trans. 1908, 93, 598-601.
- (2) Davidson, D.; Bernhard, S. A. J. Am. Chem. Soc. 1948, 70, 3426–3428.
- (3) Arnett, E. M.; Maroldo, S. G.; Schilling, S. L.; Harrelson, J. A. J. Am. Chem. Soc. 1984, 106, 6759–6767.

The Journal of Organic Chemistry

(4) (a) Lipson, V.; Gorobets, N. Mol. Diversity 2009, 13, 399-419.

(b) Dumas, A. M.; Fillion, E. Acc. Chem. Res. 2010, 43, 440-454.

(5) Hrubowchak, D. M.; Smith, F. X. Tetrahedron Lett. 1983, 24, 4951–4954.

(6) Smrcina, M.; Majer, P.; Majerova, E.; Guerassina, T. A.; Eissenstat, M. A. *Tetrahedron* **1997**, *53*, 12867–12874.

(7) Mane, R. B.; Krishna Rao, G. S. Chem. Ind. (London) 1976, 786–787.

(8) Dhuru, S. P.; Mohe, N. U.; Salunkhe, M. M. Synth. Commun. 2001, 31, 3653–3657.

(9) (a) Mignani, G.; Morel, D.; Colleuille, Y. *Tetrahedron Lett.* **1985**, 26, 6337–6340. (b) Prat, M.; Moreno-Manas, M.; Ribas, J. *Tetrahedron* **1988**, 44, 7205–7212.

(10) (a) A Meldrum's acid addition to 4-chloroquinoline has been demonstrated and presumably takes place via the N-acyl quinolinium, see: Scoville, A.; Smith, F. X. J. Heterocycl. Chem. 1977, 14, 1081–1083. (b) A Meldrum's acid addition to N-oxide quinolines has also been developed, see: Lecointre, B.; Azzouz, R.; Bischoff, L. Tetrahedron Lett. 2014, 55, 1913–1915. (c) Yousif, M. M.; Saeki, S.; Hamana, M. Chem. Pharm. Bull. 1982, 30, 1680–1691.

(11) (a) Saidi, K.; Shaterian, H. R.; Sheibani, H. Synth. Commun. 2000, 30, 2345–2351. (b) Reddy Chidipudi, S.; Khan, I.; Lam, H. W. Angew. Chem., Int. Ed. 2012, 51, 12115–12119.

(12) Chen, Z.; Jin, Y.; Stang, P. J. J. Org. Chem. 1987, 52, 4115-4117.

(13) Pinhey, J. T.; Rowe, B. A. Tetrahedron Lett. 1980, 21, 965–968.
(14) Chambers, R. D. Fluorine in Organic Chemistry; CRC Press LLC: New York, 2004; Vol. 57.

(15) For examples of addition of other activated methylenes, see: (a) Pees, K. J.; Liers, P.; Karla, C. Preparation of pentafluorophenyltriazolopyrimidines as agrochemical fungicides. Patent US5965561A, American Cyanamid Co., NJ. (b) Filler, R.; Woods, S. M. Org. Synth. **1977**, 57, 80.

(16) Kimmel, K. L.; Weaver, J. D.; Ellman, J. A. Chem. Sci. 2012, 3, 121–125.

(17) Sato, M.; Ban, H.; Kaneko, C. Tetrahedron Lett. **1997**, 38, 6689–6692.

(18) Our tentative assignment of the N-arylated product is one in which the amine has undergone arylation and dealkylation. The N-substitution occurs at the same site as the MA addition.

(19) For details, please see the Supporting Information.

(20) Substrate 2i was performed on a 1 g scale, $\sim 5\times$ the scale of 2a, and the product was obtained in an excellent yield, suggesting that the reaction may scale well.

(21) Olmstead, W. N.; Bordwell, F. G. J. Org. Chem. 1980, 45, 3299-3305.

(22) Lucius, R.; Loos, R.; Mayr, H. Angew. Chem., Int. Ed. 2002, 41, 91-95.

(23) Igumnov, S.; Don, V.; Gontar, A. Fluorine Notes 2001, 6.

(24) Fillion, E.; Dumas, A. M. J. Org. Chem. 2008, 73, 2920-2923.

(25) Helavi, V. B.; Solabannavar, S. B.; Desai, U. V.; Mane, R. B. J. Chem. Res. 2003, 2003, 174–175.

(26) (a) Ghosh, S.; De, S.; Kakde, B. N.; Bhunia, S.; Adhikary, A.; Bisai, A. Org. Lett. **2012**, *14*, 5864–5867. (b) Craig, D.; Grellepois, F. Org. Lett. **2005**, *7*, 463–465.

(27) (a) Liu, Q.; Perreault, S.; Rovis, T. J. Am. Chem. Soc. 2008, 130, 14066–14067. (b) Bélanger, G.; Lévesque, F.; Pâquet, J.; Barbe, G. J. Org. Chem. 2004, 70, 291–296.

(28) Senaweera, S. M.; Singh, A.; Weaver, J. D. J. Am. Chem. Soc. 2014, 136, 3002–3005.

(29) Fujii, S.; Maki, Y.; Kimoto, H. J. Fluorine Chem. 1989, 43, 131–144.

(30) Hansen, K. B.; Hsiao, Y.; Xu, F.; Rivera, N.; Clausen, A.; Kubryk, M.; Krska, S.; Rosner, T.; Simmons, B.; Balsells, J.; Ikemoto, N.; Sun, Y.; Spindler, F.; Malan, C.; Grabowski, E. J. J.; Armstrong, J. D. J. Am. Chem. Soc. **2009**, 131, 8798–8804.

(31) Ziegler, E.; Junek, H.; Kroboth, H. Monatsh. Chem. 1976, 107, 317–324.

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